

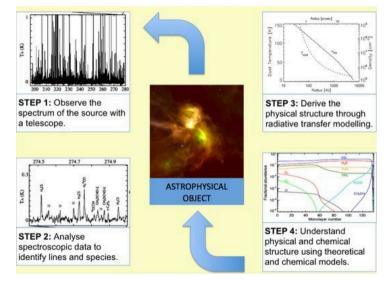
Laboratory spectroscopy of sulphur bearing species

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CAS – THE CENTER FOR ASTROCHEMICAL STUDIES





Caselli & Ceccarelli A&ARv 2012



Molecular Spectroscopy for Radioastronomy



Gas-Phase High Resolution Rotational/Ro-Vibrational Molecular Spectroscopy for Radioastronomy



Molecular Spectroscopy for Radioastronomy



Gas-Phase High Resolution Rotational/Ro-Vibrational Molecular Spectroscopy for Radioastronomy

GAS-PHASE SPECTROSCOPY AT CAS



- ► Stable (COMs) and reactive (ions and radicals) species
- ► Isotopologues characterisation
- ► THz extension of low-frequency experiments

BACK TO BASICS - SPECTROSCOPY 101



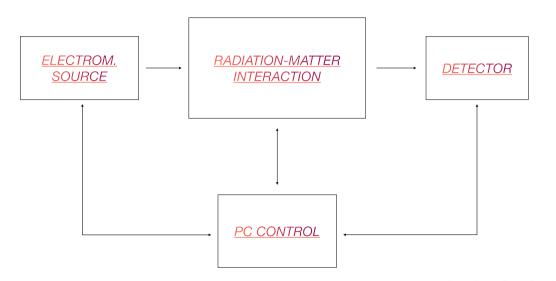
ELECTROM.
SOURCE

RADIATION-MATTER
INTERACTION

DETECTOR

BACK TO BASICS - SPECTROSCOPY 101





SPECTROSCOPY AT CAS



► SOURCE:

- CW in FM from multiplication/amplification of a cm-wave source
- Broadband Chirped-pulsed, eventually multiplied/amplified CP-FTS

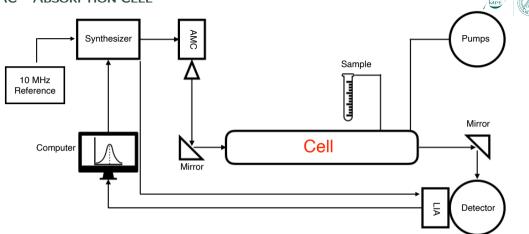
GAS RESERVOIR:

- Waveguides
- Absorption Cells <u>CASAC</u>
- Supersonic free-jet expansion chamber <u>CASJet</u>

DETECTORS:

- Room temperature Schottky diodes
- InSB Bolometers ("wet" and "dry")
- FTS heterodyne systems

CASAC - ABSORPTION CELL



Several cells are available:

- · static cell
- discharge cell
- · pyrolysis cell

<u>AMC</u>: amplifier/multiplier chain LIA: lock-in amplifier



CASAC Experiment - THEN/NOW



- ► Schottky-based active/passive multiplier chain (Virginia Diodes Inc.)
 - ▶ $80 \, \text{GHz} 1.1 \, \text{THz} \Longrightarrow \text{extended to } 1.6 \, \text{THz}$
- Three absorption cells:
 - Discharge glass tube (3m-long x 5cm-diameter)
 - Pyrolysis glass tube (3m-long x 5cm-diameter)
 - Static Cell (2.5m-long x 10cm-diameter)
- DC discharge (2kW) & solenoid (up to 350 Gauss)
- N_{2L} cooling of the cell
- Pyrolysis oven up to 1500°C
- Single and double–pass arrangement
- Cryogenic InSb (QMC) and Schottky diode (Virginia Diodes Inc.) detectors
 Dry cryogenic InSb system w/ cryocompressor
- Diffusion (VHS-6 Agilent) & mechanical (Edwards E2M40) pumps



CASAC EXPERIMENT







CASJET - MOLECULAR JET EXPERIMENT



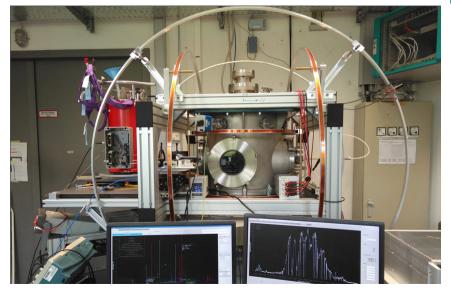
- Supersonic free-jet expansion
- ▶ Rotationally cooled molecular beam (as low as ~ 10 K)
- ▶ Pulsed valve (Series 9, Parker Hannifin), 1mm-diameter aperture
- Large diffusion + mechanical pumps (chamber pressure down to 10^{-6} – 10^{-7} Torr)
- ➤ Same source + detector as CASAC (80–1600 GHz / 4–0.2 mm)

- ► High-voltage (up to ~ 1.8 kV) DC nozzle
- ► Heating nozzle (up to ~ 300 °C)
- Stabilisation in the "zone of silence"

CASJET EXPERIMENT



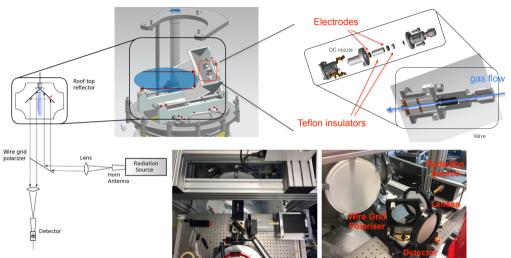




CASJET EXPERIMENT



DC-discharge Nozzle



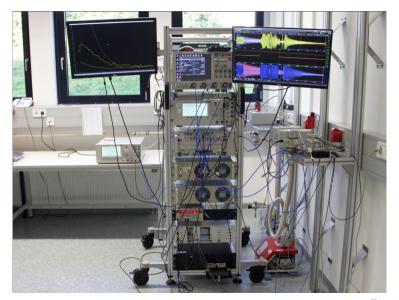


Broadband Chirped-Pulsed Fourier Transform Spectrometer

- ► 6–26 GHz (20 W and 4 W) / 75–110 GHz (250 mW) / 160–220 GHz (~2 mW)
- ▶ Detector noise figures: 6dB @ 26 GHz, 5dB @ 90 GHz, 15-20dB @ 200 GHz
- ► 5 GHz Arbitrary Waveform Generator (Keysight M8190A)
- FID recorded w/ 25 GHz scope or 2.5 GHz digitiser card
- 2 active x6 multiplier to reach 3mm band
- Mixer with sub-harmonically pumped 3mm local oscillator to reach 2mm band
- Ideal for dense and/or uncertain spectra
- Movable and flexible system

CP-FTS







Some results from the past...

PROTONATED OCS: WHAT IS KNOWN



- OCS large proton affinity (632 kJ/mol)
- ► Isoelectronic with HOCO⁺
- ▶ $N(HSCO^+)$ may be comparable with $N(HOCO^+)$

Fock & McAllister, ApJL 1982

- ► Two stable isomers: HSCO⁺ and HOCS⁺ (~21 kJ/mol higher)
 - ► HSCO⁺: $\mu_a = 1.57 \,\text{D} / \mu_b = 1.18 \,\text{D}$
 - ► HOCS⁺: $\mu_a = 1.52 \,\text{D} / \mu_b = 1.64 \,\text{D}$

Wheeler +, JCP 2006

- ▶ Higher energy isomer detected in the lab by Ohshima & Endo (CPL 1996)
- ► Laboratory HSCO⁺ detection in 2007

McCarthy & Thaddeus, JCP 2007



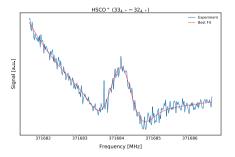
- CASAC experimental conditions:
 - ightharpoonup ~ 20 mTorr of OCS (10 %) + H₂ (10 %) + Ar
 - ► Cell at –90° *C*
 - Single-pass arrangement
 - "Anomalous" discharge conditions:

DC = 5mA / 1.5 kV and magnetic field = 200 G

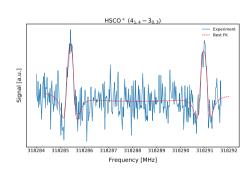
- CASJet experimental conditions:
 - 0.3% OCS in H₂
 - Valve open for 1 ms pulsing at 15 Hz
 - Discharge at 1.5 kV
 - Pressure in chamber few 10s of μTorr



CASAC



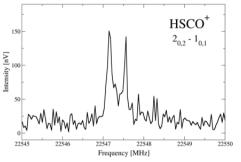
CASJet

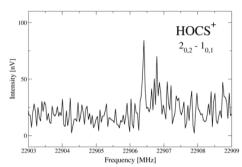


Lattanzi+, A&A 2018

CP-FTS + CASJET: FIRST TEST ON IONS







- CP-FTS coupled to CASJet
- DC @ 1.5 kV w/ 0.3% OCS in H₂
- ► $N(HSCO^+)/N(OCS) \approx 3 \times 10^{-4}$

HCSSH: DITHIOFORMIC ACID



- Sulphur analogue of formic acid, HCOOH
- ► CS₂ proposed as a major sink of sulphur on dust and ice analogues

e.g. Jiménez-Escobar+ 2014

CS₂ found in comets, both in comas (visible and UV emission) or in situ

Calmonte+ 2016

- CS₂ still undetected in interstellar ices and not detectable by radiotelescopes
- ► First detection of single substituted formic acid HC(O)SH

Rodríguez-Almeida+ 2021

► Recent detection of double-sulphur species (HS₂) opens new perspectives

Fuente+ 2017



HCSSH: EXPERIMENT



- ► Two stable isomers: trans-HCSSH and cis-HCSSH (~ 4 kJ/mol higher)
 - trans-HCSSH: $\mu_a = 1.48 \, \text{D} / \mu_b = 0.19 \, \text{D}$
 - \triangleright cis-HCSSH: $\mu_a = 2.08 \, \text{D} / \mu_b = 1.64 \, \text{D}$

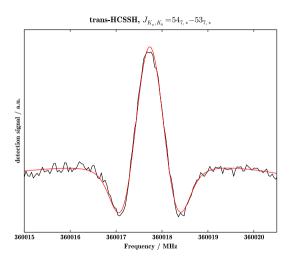
Prudenzano+ 2018

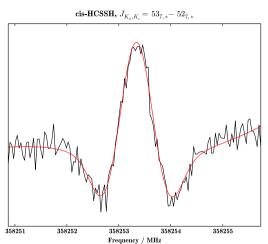
Previous cm-wave (up to 40 GHz) spectroscopy by Bak+ 1978

- CASAC experimental conditions:
 - ▶ 1:1 mixture of CS_2 and H_2 in Ar for a total pressure of 20-30 μ Torr
 - ightharpoonup DC = 40mA / 0.8 kV
 - Room temperature
- Measured 204 and 139 new transitions of trans and cis, respectively, up to 478 GHz

HCSSH: Experimental spectrum







Prudenzano+, A&A 2018



Some ongoing projects



► Thioformaldehyde detected in several astronomical regions, including singly and doubly deuterated

e.g. Marcelino+ 2005

Spectroscopically old microwave transitions available

e.g. Cox+ 1982, and Johnson+ 1971

- Database relies on astronomical observations by Marcelino+ 2005
- CDMS: "Predictions above 200 GHz should be viewed with increasing caution, especially if the calculated uncertainties exceed 0.2 MHz"

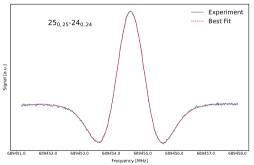
D₂CS: Experiment

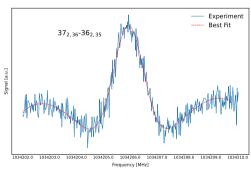


- CASAC experimental conditions:
 - ▶ 1:6:20 mixture of CS_2 and D_2 in Ar for a total pressure of \sim 30 μ Torr
 - ightharpoonup DC = 40mA / 0.65 kV
 - Cell at -25°C
- ▶ Measured more than 150 new rotational transitions in 4 days up to 1.068 THz

D₂CS: Experimental spectrum







Lattanzi+, in preparation



► Sulphur analogue of CH₃NCO, recently detected in ISM

Halfen+ 2015; Cernicharo+ 2016

► Methyl isothiocyanate (CH₃NCS), methyl thiocyanate (CH₃SCN), and, mercaptoacetonitrile (HS–CH₂–CN) are the most thermodynamically stable molecules of C₂H₃NS stoichiometry

Gronowski+ 2016

Only cm-wave spectrum (< 26 GHz) previously observed

Koput 1986

Very complex and dense spectrum, with rotational transition from the ground state and various excited torsional and vibrational bending modes

CH₃NCS: EXPERIMENT



► Very low torsional barrier (V₃=2 cm⁻¹; CH₃NCO: (V₃=21 cm⁻¹)

Koput 1986

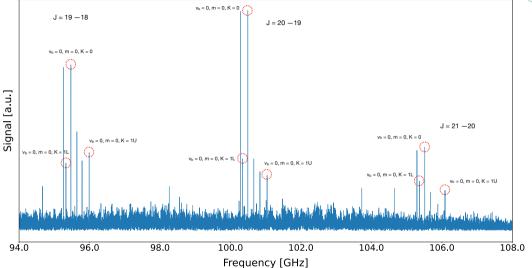
► Large dipole moment ($\mu_a = 3.4 \, \text{D}$; CH₃NCO: $\mu_a = 2.9 \, \text{D}$)

Lett & Flygare 1967 1986

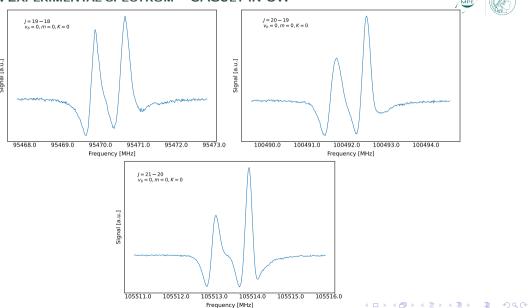
- CASJet experimental conditions:
 - ► Solid sample: melting point ~ 37°C
 - ▶ Bubbler at ~80°C with He flow
 - no DC
- Measured several lines in the 35-125 GHz with a combination of CW in FM and CP-FTS
- ▶ Preliminary fit of lower K's available, but **very much work in progress**
- ▶ New measurements w/ Heating Nozzle to come "soon"...

CH3NCS: Experimental spectrum – CASJet + CP-FTS





CH3NCS: Experimental spectrum – CASJet in CW





Next?

CURRENT PROJECTS



- Protonated Sulphur Monoxide: HSO⁺
 - Large dipole moment: $\mu_a = 2.87 \, \text{D} / \mu_b = 1.07 \, \text{D}$
 - Closed-shell molecule: favourable partition function compared to SO
 - Predictions in hand; few scans already performed; testing some candidate lines

- Protonated thioformaldehyde: H₂CSH⁺
 - ▶ Very large dipole moment: $\mu_a = 4.64 \, \text{D} / \mu_b = 1.96 \, \text{D}$
 - ab initio predictions in hand; experiment to start in the next weeks

CONCLUSIONS



- Sulphur chemistry still largely to discover
- Sulphur COMs, including isotopic substituted and isomeric forms
- Not always straightforward to pass from Oxygen to Sulphur
- First double-S in ISM opens up new scenarios