



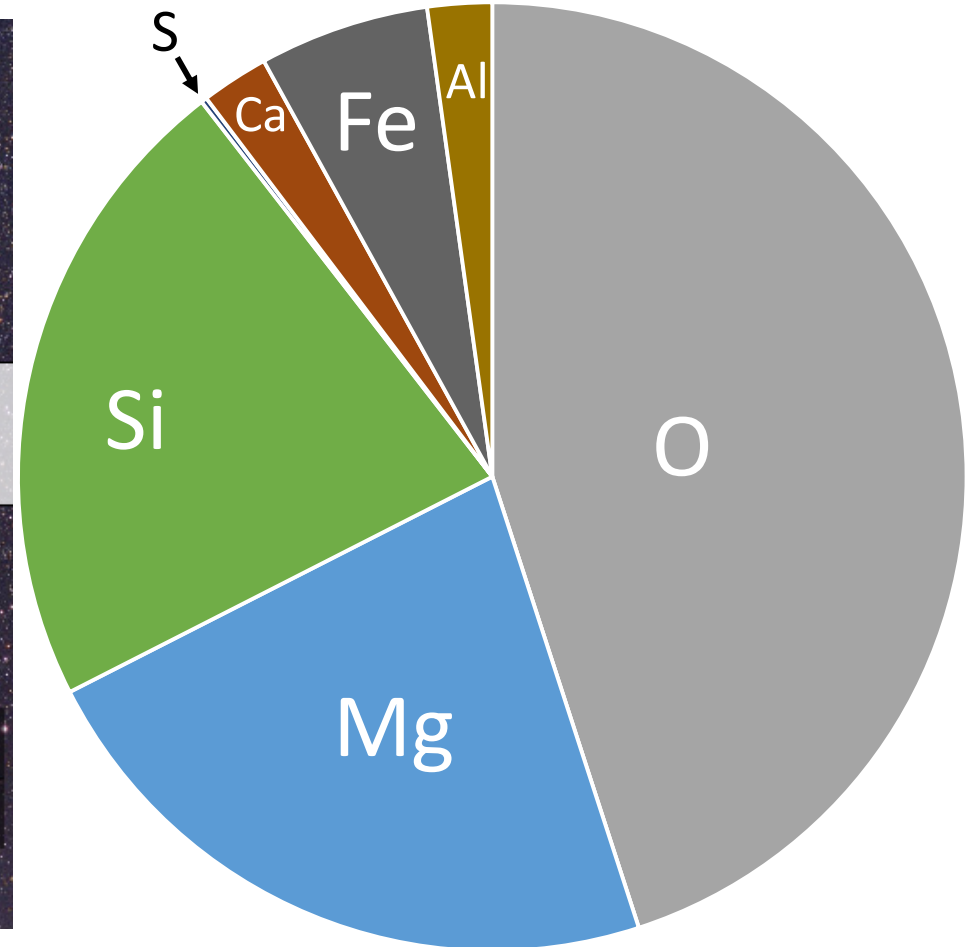
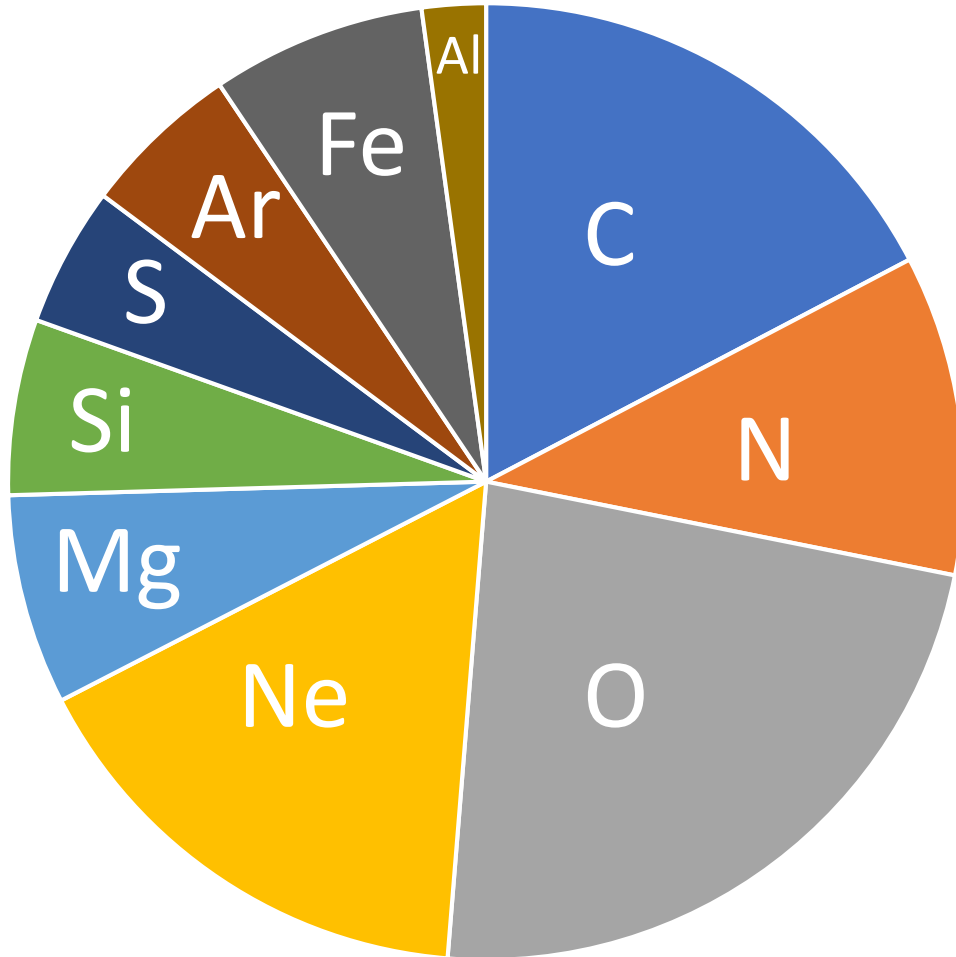
Astrochemistry Does Not (Always) Need Carbon

Ryan C. Fortenberry, Ph.D.
Assoc. Prof. of Chemistry
University of Mississippi

III Chilean Astrochemistry School, November 29, 2022



Elemental Abundance



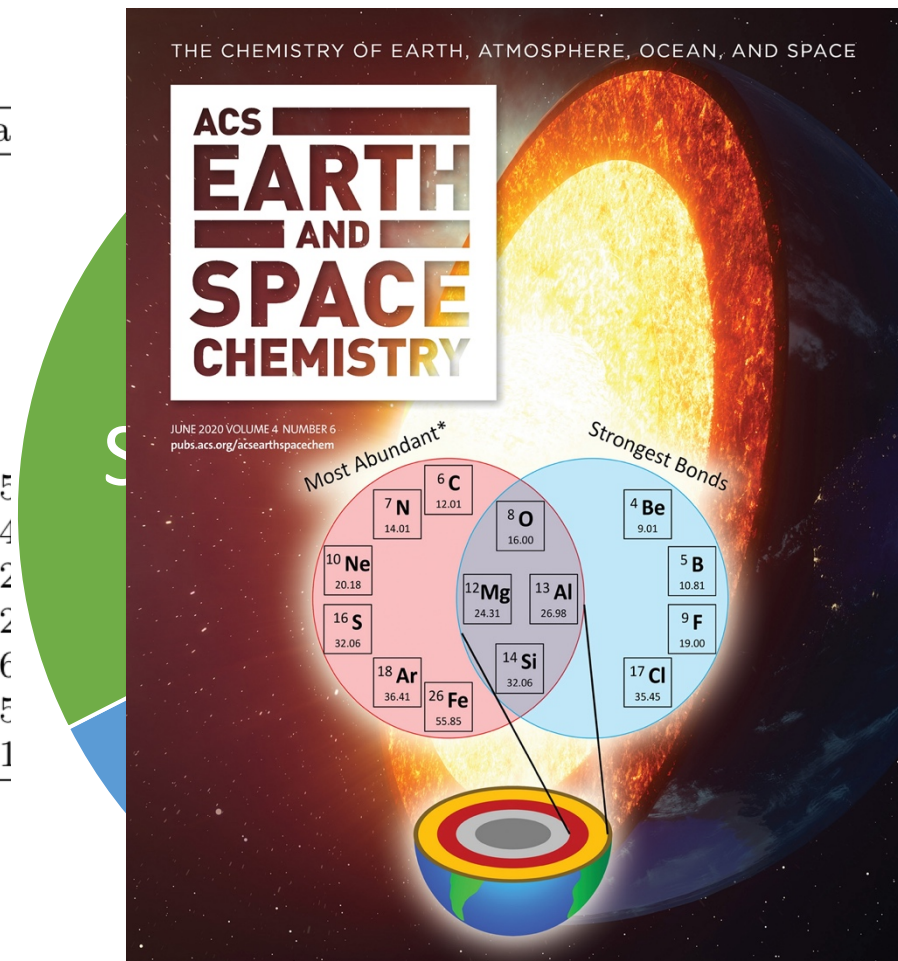


H_nX-YH_m Bond Strengths

	Li	Be	B	C	N	O	F	Na
Li	-23.54							
Be	-40.99	-69.54						
B	-42.78	-79.75	-100.55					
C	-44.62	-89.26	-99.66	-83.70				
N	-69.41	-117.18	-135.43	-77.80	-58.99			
O	-99.41	-143.52	-146.01	-85.50	-71.89	-43.38		
F	-134.20	-152.70	-164.80	-104.43	-63.59	-42.63	-32.17	
Na	-19.61	-32.06	-31.79	-48.97	-76.50	-111.08	-16.50	
Mg	-33.87	-53.00		-82.34	-109.34	-141.50	-28.40	
Al	-40.53	-63.00	-72.97	-105.34	-126.32	-156.00	-33.20	
Si	-48.13	-73.22	-80.87	-82.00	-97.10	-118.00	-146.28	
P	-49.56	-75.41	-83.18	-65.37	-91.00	-83.32	-105.75	
S	-77.03	-102.30	-107.82	-68.43	-60.00	-63.91	-77.32	
Cl	-108.69	-130.25	-119.83	-77.75	-55.19	-50.04	-55.62	-95.10

kcal/mol

CCSD(T)-F12b/cc-pVTZ-F12

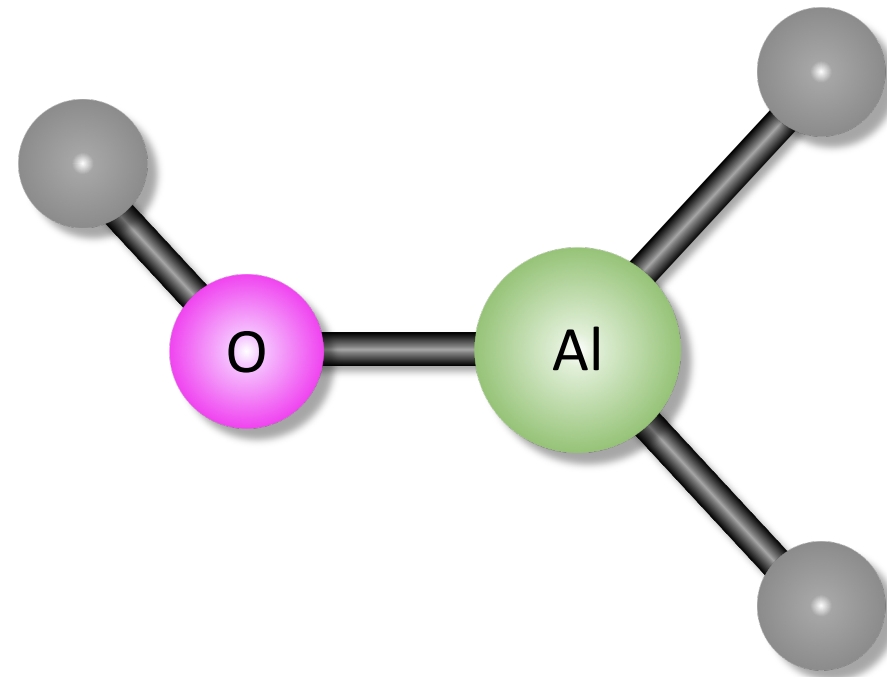
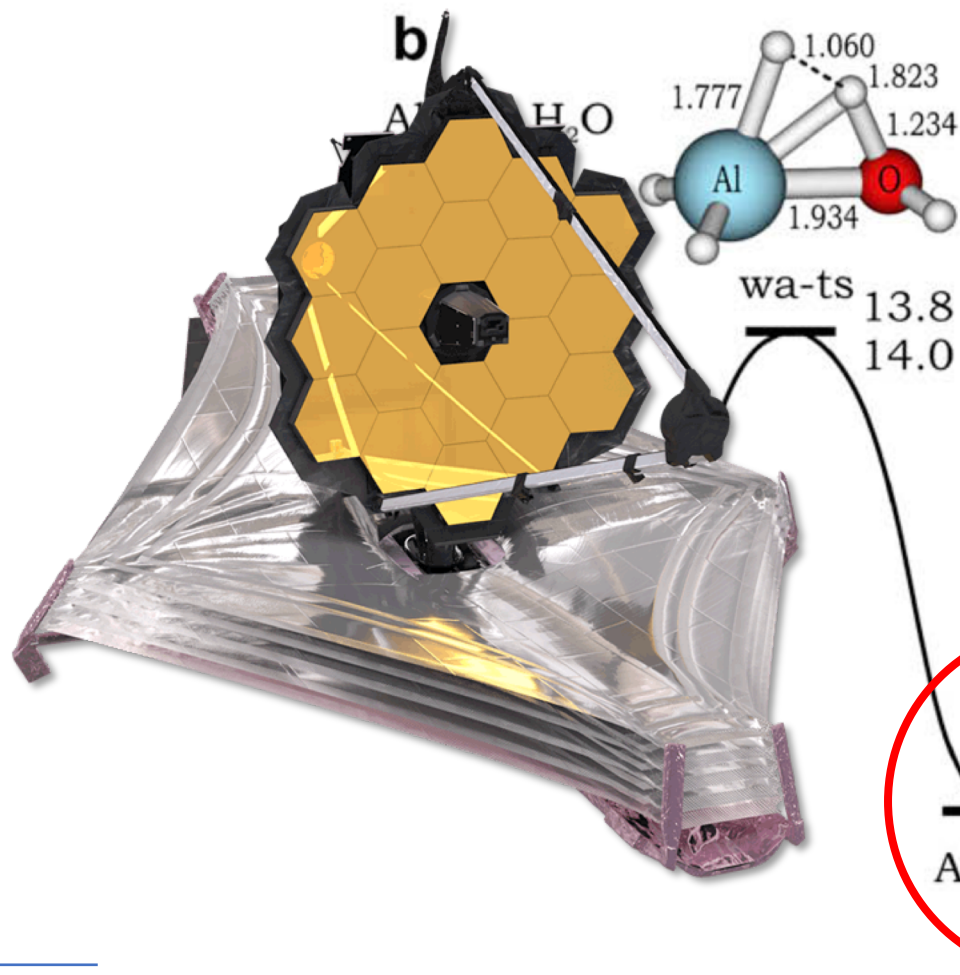


Cl

51.79



H_nX-YH_m Molecules





Computing Anharmonic Freqs

$$H = \frac{1}{2} \sum_{\alpha\beta} (J_\alpha - \pi_\alpha) \mu_{\alpha\beta} (J_\beta - \pi_\beta) - \frac{1}{2} \sum_k \frac{\partial^2}{\partial Q_k^2} - \frac{1}{8} \sum_\alpha \mu_{\alpha\alpha} + V(\mathbf{Q})$$

$$V = \frac{1}{2} \sum_{ij} F_{ij} \Delta_i \Delta_j + \frac{1}{6} \sum_{ijk} F_{ikj} \Delta_i \Delta_j \Delta_k + \frac{1}{24} \sum_{ijkl} F_{ikjl} \Delta_i \Delta_j \Delta_k \Delta_l$$

Geometry
Optimization
(F12-TZ)

Generate
Points
(0.005 Å or
0.005 rad.)

Calculate
Energies
(F12-TZ)

Least-
Squares Fit

Transform
Coordinates

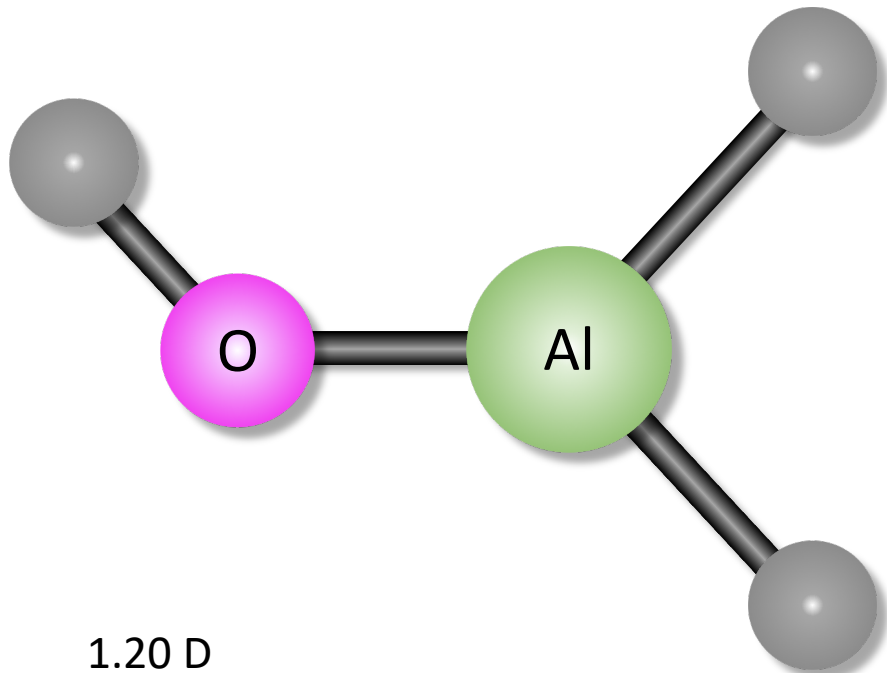
Compute
Frequencies
& Rotational
Constants
(VPT2)

X. Huang & T. J. Lee, *J. Chem. Phys.* **129**, 044312 (2008).

R. C. Fortenberry & T. J. Lee, *Ann. Rep. Comput. Chem.* **15**, 173-202 (2019).



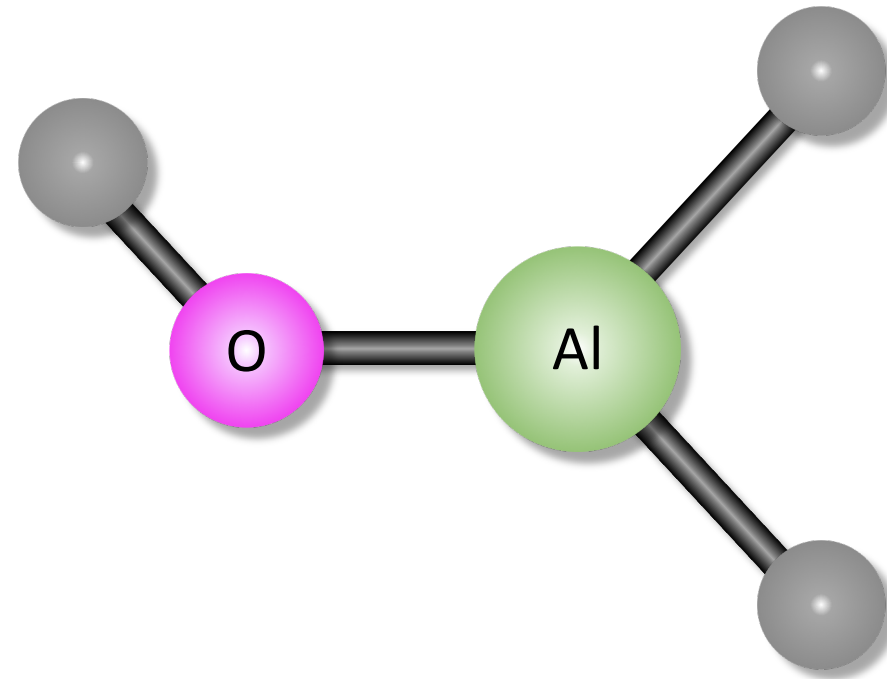
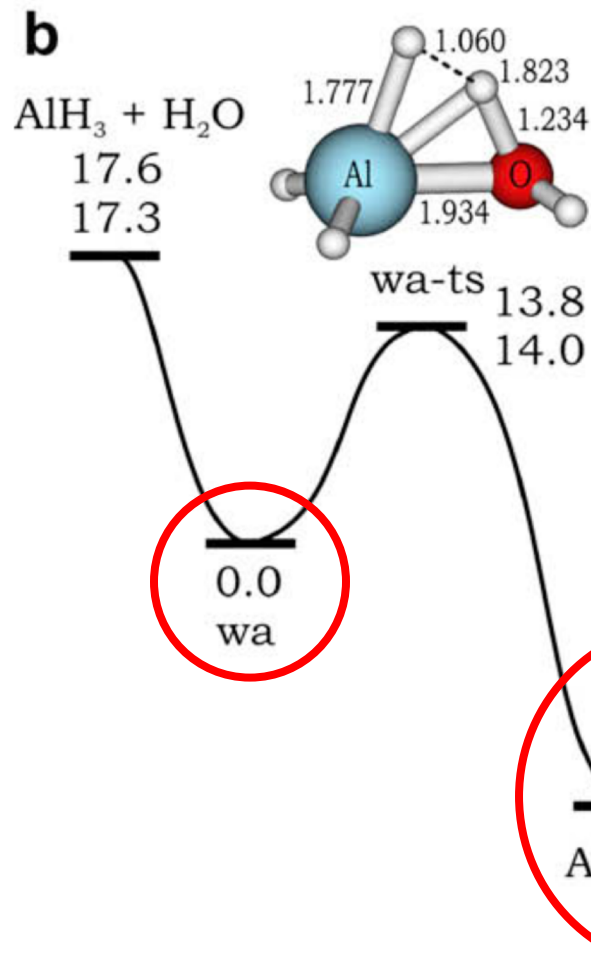
H_nX-YH_m Molecules



AlH_2OH	f	F12-TZ
$\nu_1 \text{ cm}^{-1}$	83	3768.7
$\nu_2 \text{ cm}^{-1}$	170	1918.4
$\nu_3 \text{ cm}^{-1}$	179	1892.5
$\nu_4 \text{ cm}^{-1}$	203	855.7
$\nu_5 \text{ cm}^{-1}$	76	766.3
$\nu_6 \text{ cm}^{-1}$	260	621.9
$\nu_7 \text{ cm}^{-1}$	218	616.1
$\nu_8 \text{ cm}^{-1}$	84	486.3
$\nu_9 \text{ cm}^{-1}$	104	424.1

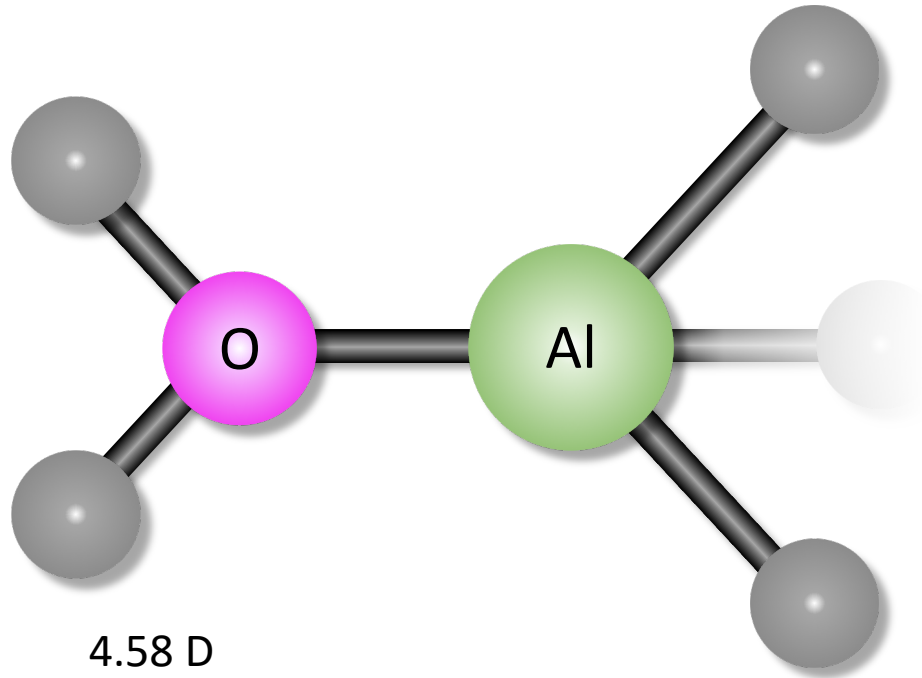


H_nX-YH_m Molecules





AlH₃OH₂

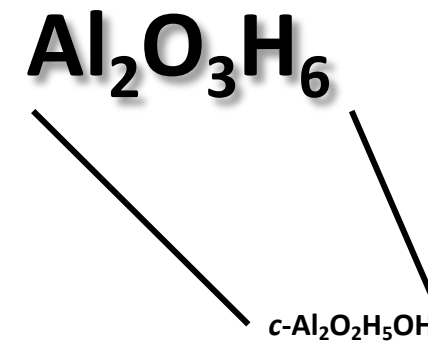


	<i>f</i>	F12-TZ
$\nu_1 \text{ cm}^{-1}$	142	3715.0
$\nu_2 \text{ cm}^{-1}$	62	3624.5
$\nu_3 \text{ cm}^{-1}$	168	1854.5
$\nu_4 \text{ cm}^{-1}$	194	1819.3
$\nu_5 \text{ cm}^{-1}$	367	1809.0
$\nu_6 \text{ cm}^{-1}$	91	1607.4
$\nu_7 \text{ cm}^{-1}$	261	785.1
$\nu_8 \text{ cm}^{-1}$	297	778.9
$\nu_9 \text{ cm}^{-1}$	412	775.3
$\nu_{10} \text{ cm}^{-1}$	5	664.0
$\nu_{11} \text{ cm}^{-1}$	153	487.9
$\nu_{12} \text{ cm}^{-1}$	148	319.7





H_nX-YH_m Molecules





H_nX-YH_m Bond Strengths

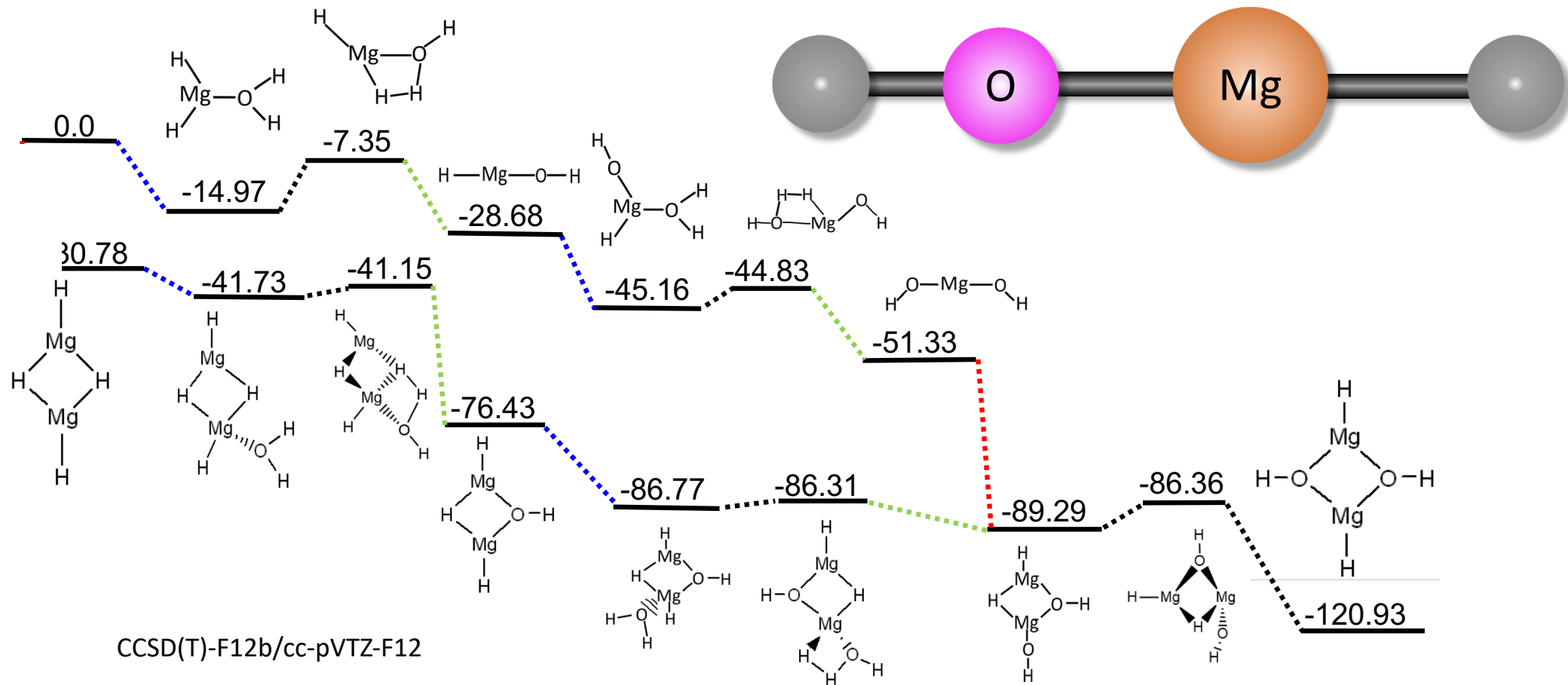
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P	-49.56	-75.41	-83.18	-65.37	-69.91	-83.32	-105.75	-39.62	-57.80	-67.13	-65.44	-51.26		
S	-77.03	-102.30	-107.82	-68.43	-60.33	-63.91	-77.32	-64.50	-82.61	-92.54	-82.21	-60.19	-57.55	
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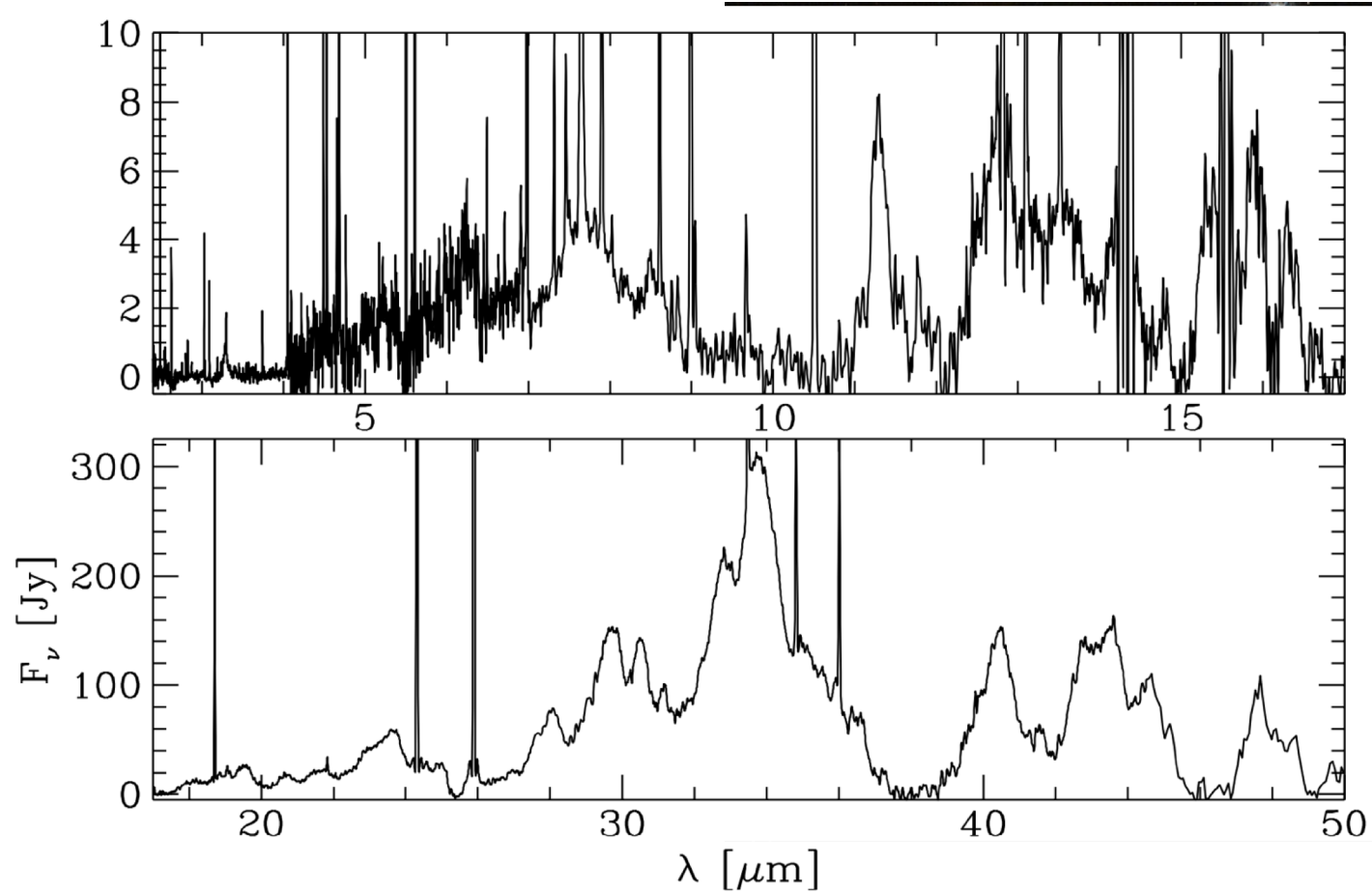
kcal/mol

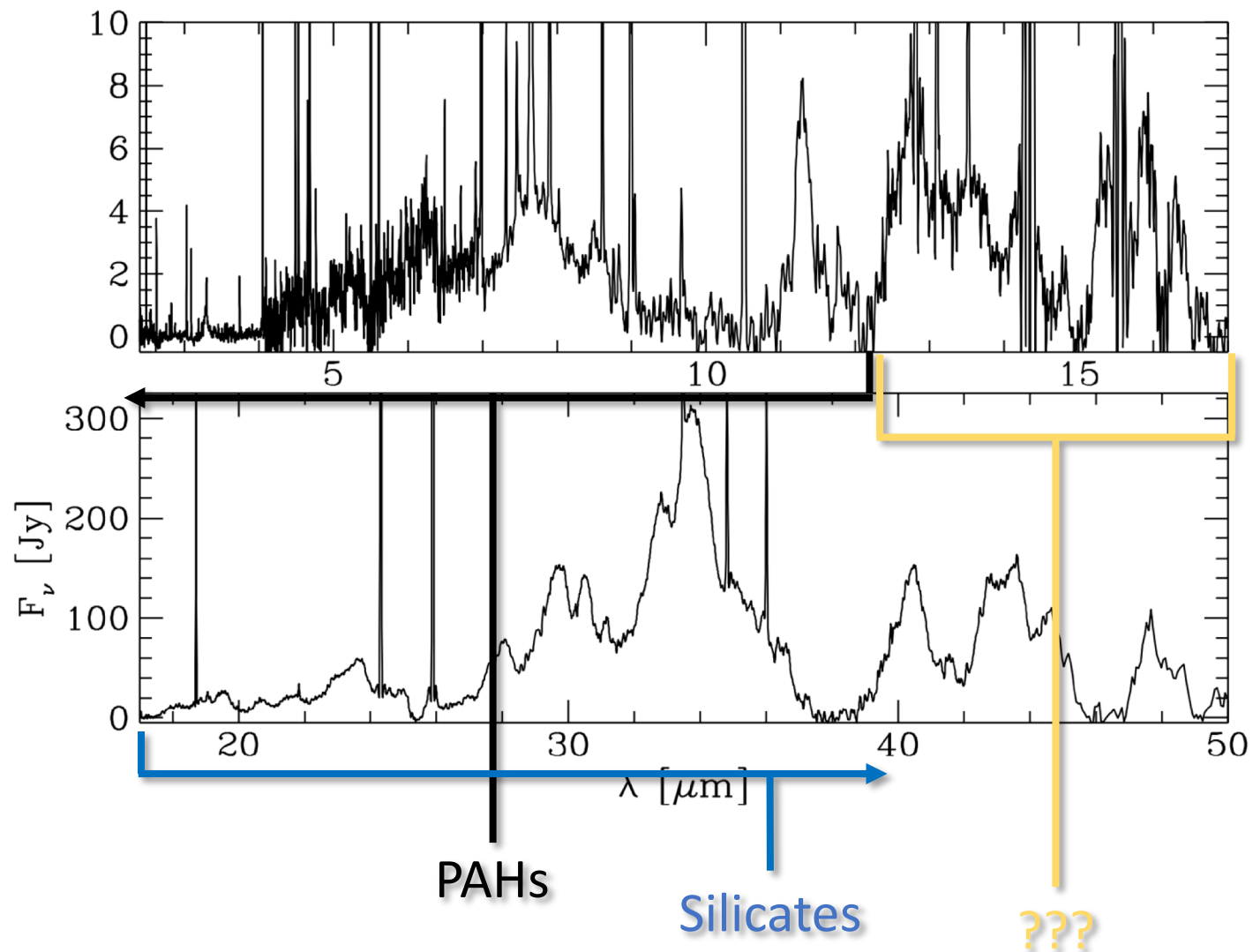
CCSD(T)-F12b/cc-pVTZ-F12



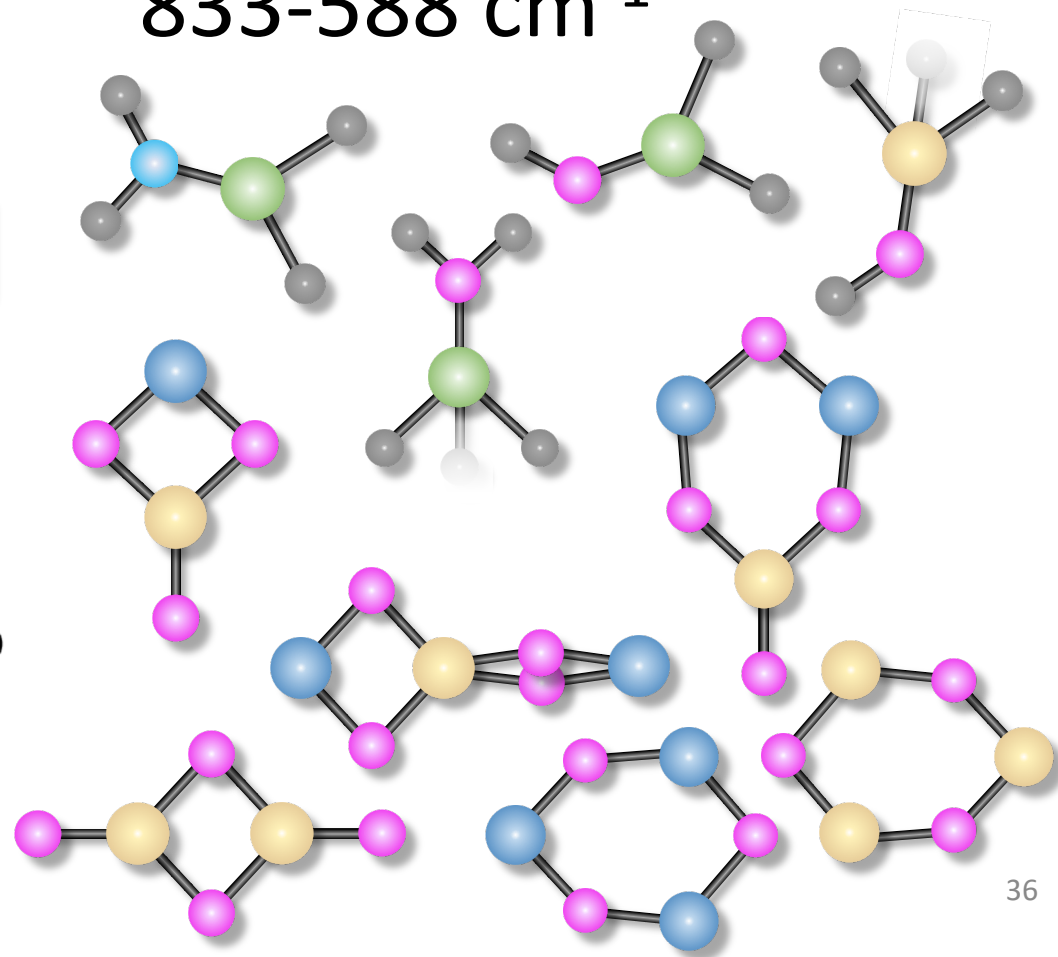
H_nX-YH_m Molecules







12-17 μm
833-588 cm^{-1}





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